

Producing Olefinic and Fuel (Cont.)

SOV/3734

Gasification installations for producing olefinic gases (ethylene, propylene)

207

The Catarol process	208
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AVAILABLE: Library of Congress

Card 6/6

JA/gmp
6-26-60

VISHNEYAKOVA, L.P.

PHASE I BOOK EXPLOITATION

SOT/941

Mezhvirskoye sovetskoye po khimi i nafti, Moscow, 1956.

Sbornik trudov Mezhvirskogo sovetschaniya po khimi i nafti
(Collection of Transactions of the Inter-University Con-
ference on Petroleum Chemistry) [Moscow] Izd-vo Nauk.
univ., 1956. 313 p. Arata slip inserted. 1,600 copies
printed.

Organizing Committee of the Conference: Chairman: B. A.

Kazanskiy, Academician; Vice-Chairman: S. I. Kurnakov,

Professor; Secretary: Ye. S. Biletnikov, Professor; A. P. Plate, Prof-

essor; Scientific Worker: V. V. Gostun-

skaya, T. M. Titova-Skorobogatova, L. A. Krivankaya.

PURPOSE: This collection of articles is intended for the
teaching staff of universities and schools on higher ed-
ucation training specialists for the petroleum and petroli-

um-refining industries.

CONTENTS: The collection includes articles dealing with the
present state of the petroleum industry, the scientific
research problems in petroleum chemistry, the chemistry
of petroleum, the composition of petroleum and petroleum
products, the scientific principles of refining petroleum
into motor fuels and lubricants and the manufacture of
synthetic products from hydrocarbon gases and petroleum.
One article discusses the effect of chemical composition
and additives on fuel combustion in jet engines. Material
on Petroleum Chemistry held at the Moscow State University
by Iuemi M. V. Lomonosov November 26-30, 1956. No per-
sonalities are mentioned. References accompany most of the
articles.

TABLE OF CONTENTS: None given

The authors and the titles of articles are as follows:

Introduction by B. A. Kazanskiy, Academician

Card 2/7

Collection of Transactions (Cont.)

SOV/4941

Kiselev, A. V., Laboratoriya adsorbsii Moskovskogo gosudarstvennogo universiteta (Adsorption Laboratory of the Moscow State University) and Laboratoriya sorbtionnykh protsessov Instituta fizicheskoy khimii AN SSSR (Laboratory of Sorption Processes, Institute of Physical Chemistry, AS USSR). Hydrocarbon Adsorption Energy 258

Paushkin, Ya. M., R. V. Sychev, T. P. Vishnyakova, and A. K. Zhomov, Moscow Petroleum Institute imeni I. M. Gubkin. Effect of Chemical Composition and Additives on Fuel Combustion in Jet Engines 293

AVAILABLE: Library of Congress (TP690.A1M445 1956)

Card 7/7

JA/wrc/ec
4-20-61

S/152/60/000/009/003/004/XX
B024/B076

AUTHORS: Zhomov A. K., Vishnyakova T. P., and Paushkin Ya. M.

TITLE: Kinetics of High-Temperature Pyrolysis of Crude Oil
to Gas With a High Olefin Content

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy. Neft' i gaz,
1960, No. 9, pp. 103 - 107

TEXT: The authors consider the possibility of applying
G.M. Panchenkov's theory on the kinetics of thermal cracking of
petroleum hydrocarbons to the description of the pyrolysis of crude
oil residues in the presence of steam. In cooperation with
V. S. Tret'yakova (Ref. 3) G. M. Panchenkov obtained an equation
from which the velocity constants of the first and second stages of
a continuous first-order reaction in the cracking process can be
determined:

Card 1/3

Kinetics of High-Temperature Pyrolysis
of Crude Oil to Gas With a High Olefin
Content

S/152/60/000/009/003/004/xx
B024/B076

$$n_0 \frac{dx}{dl} = \frac{k_1 (1-x)}{v_2^x + \frac{v_5}{v_2} \left[v_3^x - \frac{v_3(1-x)}{1-k} - v_3 \frac{(1-x)^k}{1-k} \right]} , \quad (2)$$

✓
—

where x denotes the degree of conversion; l the distance from the beginning of the reaction zone; v_1, v_2, v_3, v_5 are the stoichiometric coefficients; n_0 is the number of gram-moles of the initial cracking residue; k, k_1 are the reaction constants. By means of a graphic solution of this transformed equation the authors ascertained that the equation obtained for thermal cracking is also applicable to high-temperature pyrolysis. There are 4 figures and 5 Soviet references.

Card 2/3

Kinetics of High-Temperature Pyrolysis
of Crude Oil to Gas With a High Olefin
Content

S/152/60/000/009/003/004/xx
B024/B076

ASSOCIATION: Moskovskiy institut neftekhimicheskoy i gazovoy
promyshlennosti im. akad. I. M. Gubkina
(Moscow Institute of the Petrochemical and Gas
Industry imeni Academician I. M. Gubkin)

SUBMITTED: January 29, 1961

✓

Card 3/3

PAUSHKIN, Ya.M.; VISHNYAKOVA, T.P.; CHERNUKHINA, V.G.

Catalytic reforming of naphthenic hydrocarbons of gasoline fractions
into aromatic hydrocarbons on a catalyst with 0.1-0. 3% of nickel.
Izv. vys. ucheb. zav.; neft' i gaz 4 no.5:69-73 '61. (MIRA 15:2)

1. Moskovskiy institut neftekhimicheskoy i gazovoy promyshlennosti
imeni akad.I.M.Gubkina.
(Hydrocarbons) (Cracking process) (Catalysts, Nickel)

53300

2109 only

23486
S/152/61/000/005/001/002
B126/B219

AUTHORS: Paushkin, Ya. M., Vishnyakova, T. P., and Chernukhina, V. G.

TITLE: Catalytic reforming of naphthenic hydrocarbons to aromatic hydrocarbons from benzine fractions using a catalyst with 0.1 - 0.3% nickel

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy. Neft' i gaz, no. 5, 1961, 69 - 73

TEXT: For petrochemical synthesis the problem of aromatic hydrocarbons obtaining from crude oil is of current importance. The dehydrating effect of nickel catalysts has already been carefully examined by A. D. Zelinskiy and his school. Ciapetta (Ref. 2, Ciapetta F., Hanter I., Ind. Eng. Chem., 45, 147, 1953) showed that isomerization of normal pentane, hexane, heptane, and octane to isoparaffins is possible with a catalyst containing 5% of nickel on aluminum silicate and at 407°C, 25 atm pressure; (yield 55 - 65%). Kh. M. Minayev, N. I. Shuykin, L. M. Feofanova and Yu. P. Yegorov isomerized normal decane and hendecane with a catalyst containing 8% of nickel on aluminum oxide. The authors

Card 1/6

23486
S/152/61/000/005/001/002
B126/B219

Catalytic reforming of...

of the present paper experimented with nickel catalysts containing 0.1 - 0.3% of nickel on aluminum oxide. The catalyst was prepared from the active form of aluminum oxide, obtained by calcining ordinary aluminum oxide at 700°C, whereupon the γ -form Al_2O_3 is achieved. The aluminum oxide obtained was soaked with a nickel nitrate solution of $\text{Ni}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ in such quantities as to obtain the necessary concentration of metallic nickel on Al_2O_3 after evaporation. The best experimental results were obtained with catalysts containing 0.1 to 0.3% of nickel. They are given in Tables 3 and 4. A catalyst with 0.1 - 0.3% of Ni on Al_2O_3 works without any noticeable decrease in activity for 10 - 12 hr at a volume rate of 0.2 hr^{-1} , then the activity drops as a result of coking. Regeneration was effected by burning the coke at 400 - 500°C. In Table 5, a comparison between reforming by nickel and reforming by platinum is given. The experiments thus proved that a catalyst on a nickel basis only differs slightly in its activity from a catalyst on Pt-basis, but it is much cheaper. There are 5 tables and 3 references: 2 Soviet-bloc and 1 non-Soviet-bloc. The reference to the English-language publication reads as follows: Ciapetta F., Hanter I., Ind. Eng. Chem., 45, 147, 1953.

Card 2/6

Catalytic reforming of...

23486
S/152/61/000/005/001/002
B126/B219

ASSOCIATION: Moskovskiy institut neftekhimicheskoy i gazovoy
promyshlennosti im. akad. I. M. Gubkina (Moscow Institute of
Petrochemical and Gas Industry imeni Acad. I. M. Gubkin)

SUBMITTED: February 26, 1961

1) Показатели	2) Температура опыта, °C			3) 0,1% Ni на Al ₂ O ₃ при 550°
	450	500	550	
4) Плотность D ₄ ²⁰	0,7360	0,751	0,768	0,782
5) Молекулярный вес	110,5	118	128	139
6) Броминное число	5,5	10,5	13	10,2
7) Групповой состав, % вес:				
7) прометические углеводороды	6,1	13,3	21,5	31,1
8) изфтеповые	--	36,7	30	20,2
9) парафиновые	--	42,3	38	39,8
10) непредельные	3,9	7,7	10,5	8,9
11) Состав газа (% объемн.)				
12) водород	58	73	66,7	70-80
13) непредельные	3,3	7,5	8,8	—

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Table 3

S/065/61/000/012/003/005
E075/E135

AUTHORS: Vishnyakova, T.P., Paushkin, Ya.M., Bondarenko, L.V.,
and Smirnov, A.P.

TITLE: Influence of the chemical composition of hydrocarbon
feedstock and aqueous vapours on the dynamics of
formation of olefines during high temperature pyrolysis

PERIODICAL: Khimiya i tekhnologiya topliv i masel, no.12, 1961.
11-14

TEXT: The aim of this work was to study dynamics of
gasification of n-cetane, α -methyldecalin and a middle kerosene
fractions (b.pt.200-300 $^{\circ}$ C) leading to the formation of ethylene
and propylene. The gasification process was carried out in a
laboratory apparatus, a diagram of which is shown in Fig.1, where:
1 - reactor; 2 - electric furnace; 3 - flow meters; 4 - receiver
for condensate; 5 - water pump; 6 - feedstock pump; 7 - burettes;
8 - receiver for condensate; 9 - condenser; 10 - water washer;
11 - oil washer; 12 - gas meter; 13 - beater for feedstock;
14 - heater for steam; 15 - sprayer. The feedstock was preheated
to 300 $^{\circ}$ C, sprayed into the reactor with steam preheated to

Card 1/43

Influence of the chemical

S/065/61/000/012/003/005
E075/E135

450-500 °C (feedstock-steam ratio 1:1). The mixture was heated in the reactor to 800 °C, the temperature being controlled electrically. The total material balance and the balance for each section of the reactor are obtained as a function of the place of gas take-off. The time of contact of feedstock in the reaction zone was determined to obtain the speed of gasification of the different types of hydrocarbons along the length of the reactor. For the n-cetane fraction the formation of olefines passes through a maximum and reaches about 40% of the total gas for the reaction times of 0.5 to 0.6 sec. Subsequently the concentration of olefines begins to fall rapidly and for 1.5 - 2.0 sec reaction times it is as low as 5-7%. The extent of gasification after 2 sec reaches 90% of the feedstock but at the time of maximum olefine yield, only 50% of the feedstock is gasified. Gasification of α -methyldecalin fraction gives less olefines and a maximum yield of 24% is reached for the reaction time of 0.6 sec. The kerosene fraction, which consisted mainly of naphthenes and paraffins, gave a maximum yield of 27% after 0.3-0.5 sec. The composition of gases formed during the pyrolysis is different for each hydrocarbon fraction investigated.

Card 2143

Influence of the chemical

S/065/61/000/012/003/005;
E075/E135

There are 4 figures and 1 table.

ASSOCIATION: MINKh and GP imeni I.M. Gubkin

Card 3/43

VISHNYAKOVA, T.P.; PAUSHKIN, Ya.M.; BONDARENKO, L.V.; SMIRNOV, A.P.

Effect of the chemical composition of hydrocarbon raw charge
and water vapor on the dynamics of olefin formation at high
temperature pyrolysis. Khim.i tekhnopl.i masel 6 no.12:11-
14 D '61. (MIRA 15:1)

1. Moskovskiy institut neftekhimicheskoy i gazovoy promyshlennosti
im. akad. Gubkina.
(Hydrocarbons) (Olefins)

PAUSHKIN, YA.M., POLAK, L.S., VISHNYAKOVA, T.P., PATALAKH, I.I.,
MACHUS, F.F., SOKOLINSKAYA, T.A.

New ferrus-containing polymers on the basis of ferrocene and their electrophysical
properties.

Report submitted for the International Symposium of Macromolecular chemistry
Paris, 1-6 July 63

~~compound, knock inhibitor, monomer, manganese containing~~
~~A~~STRACT: The production of cyclopentadiene and its homologs is of great importance

"APPROVED FOR RELEASE: 09/01/2001

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merization and high accuracy

Cont. 1/3

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"APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001860110008-6

ACCESSION NR. AT4008695

FIR spectra which indicates the presence of 2 unpaired electrons. The electro-

APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001860110008-6"

PAUSHKIN, Ya.M.; VISHNYAKOVA, T.P.; SMIRNOV, A.P.; ANAN'YEV, P.G.;
NEPRYAKHINA, A.V.

Recent developments in the cracking of hydrocarbons; cracking
with heat given off and cracking cut off at high temperatures.
Trudy MINKHIGP no.44:118-128 '63. (MIRA 18:5)

L 16996-63
RM/NW/JD/JW/MAY

EWP(j)/EPF(c)/EWP(q)/EWT(m)/BDS AFFTC/ASD PC-4/PR-4

S/204/63/003/002/006/006 77
76

AUTHOR: Paushkin, Ya. M., Makarova, T. P., Sokolinskaya, T. A., Zimina,
K. I., and Kotova, G. G.

TITLE: Alkylation of Ferrocene by olefins in the presence of the com-
pounds boron fluoride and aluminum chloride

PERIODICAL: Neftekhimiya, v. 3, no. 2, 1963, 280-284

TEXT: The number of olefins used for alkylation of ferrocene was ex-
panded, and such catalysts as the strong complex acid $H_3PO_4 \cdot BF_3$ and $BF_3 \cdot O(C_2H_5)_2$, in addition to $AlCl_3$, were used, which allowed the concept on the
mechanisms of ferrocene alkylation to be widened and new previously unknown
alkylferrocenes to be synthesized. A considerable increase in yields of mono-
alkyl derivatives of ferrocene is reported. The mono- and di-isooctylferro-
cenes were obtained by the direct alkylation of ferrocene by olefins. There
are 3 tables and 1 figure. The most important English-language references read
as follows: T. I. Kealy, P. L. Pavson, Nature, 168, 1039, 1951; G. Wilkinson,
F. A. Cotton, J. M. Birmingham, J. Inorg. and Nucl. Chem., 295, 1956.

ASSOCIATION: Moscow Institute of Petrochemical and Gas Industry imeni I.M. Gubkin.
Card 1/2,

PAUSHKIN, Ya.M.; VISHNIAKOVA, T.P.; PATALAKH, I.I.; SOKOLINSKAYA, T.A.;
MACHUS, F.F.

Ferrocene-based synthesis of polymers and some of their electro-
physical properties. Dokl. AN SSSR 149 no.4:856-859 Ap '63.
(MIRA 16:3)

1. Institut neftekhimicheskoy i gazovoy promyshlennosti im. I.M.
Gubkina. Predstavлено академиком A.V.Topchiyevym.
(Polymers) (Ferrocene)

VISHNYAKOVA, T.P.; PAUSHKIN, Ya.M.; KLIMENKO, M.Ya.; MAR'YASHKIN, N.Ya.

Oxidation of η -butylenes to methyl ethyl ketone in the presence of
a palladium chloride catalyst. Izv.vys.ucheb.zav.; khim.i khim.tekh.
7 no.6:989-992 '64. (MIRA 18:5)

1. Moskovskiy institut neftekhimicheskoy i gazovoy promyshlennosti
imeni Gubkina, kafedra neftekhimicheskogo sinteza.

"APPROVED FOR RELEASE: 09/01/2001

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APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001860110008-6"

"APPROVED FOR RELEASE: 09/01/2001

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CIA-RDP86-00513R001860110008-6

APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001860110008-6"

ACCESSION NR: AP4043278

ASSOCIATION: MINKh and GP

SUBMITFD: 00

ENCL: 00

SUB CODE: GC, TD, N, A, S, V, C, P, R, E, F, H, I, J, K, L, M, N, O, P, Q, R, S, T, U, V, W, X, Y, Z

Card 2/2

"APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001860110008-6

TITLE: Preparation of polynitrogen compounds by the reaction of calcium carbide with atomic oxygen

ABSTRACT: A method for the preparation of polynitrogen compounds by the reaction of calcium carbide with atomic oxygen.

APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001860110008-6"

ACCESSION NR: AP4045016

conjugated polymers. The carbonyl compounds—acetone, acetophenone,
acraldehyde, and acetylacetone—reacted with calcium chloride in
molar ratios of 1 mole to 1 mole, respectively. The polymer influences
the rate of reaction.

and 215

"APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001860110008-6

Was synthesized for the first time. Most of the available evidence
now indicates that it was made in the United States by the Soviet Union.

This was a very explosive explosive developed at Sverdlovsk and had a molecular

STRUCTURE: 1,3,2,4,5-PENTANEPHOSPHORIC ACID, MONOPOTASSIUM SALT

APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001860110008-6"

ACCESSION NR: AP4030375

S/0190/64/006/003/0545/0550

AUTHOR: Paushkin, Ya. M.; Polak, L. S.; Vishnyakova, T. P.;
Patalakh, I. I.; Machus, F. F.; Sokolinskaya, T. A.

TITLE: New iron-containing ferrocene-based polymers and their
electrophysical properties.

SOURCE: Vy*skomolekulyarny*ye soyedineniya, v. 6, no. 3, 1964, 545-
550

TOPIC TAGS: organic semiconductor, semiconducting polymer, ferrocene
polymer, ferrocene polymer preparation, electrical property

ABSTRACT: Fourteen new polymers based on ferrocene and a number
of aromatic compounds have been prepared by polyrecombination or
polycondensation, and their electrical properties have been studied
at the Moscow Institute of the Petrochemical and Gas Industry imeni
Gubkin. The polyrecombination of ferrocene and α -bromonaphthalene,
 p -dichlorobenzene, benzonitrile, salicylic acid, salicylaldehyde, or
benzaldehyde, and of isobutyl-, isopentyl-, or isoctylferrocene alone

Card 1/3

ACCESSION NR: AP4030375

was carried out at 175-200°C in the presence of tert-butyl peroxide at various starting material-to-peroxide molar ratios. Yields of 3-39% for soluble (dark-brown) and 23-77% for insoluble (black) solid polymers were obtained. The polycondensation of ferrocene with acetone in the presence of ZnCl₂ and hydrogen chloride at 56°C formed soluble polymers; that of acetyl- or 1,1'-diacetylferrocene alone in the presence of ZnCl₂ at 200°C and 180°C respectively yielded both soluble and insoluble polymers. All the polymers but alkylferrocene-polyrecombination products gave a one-component signal in the EPR spectrum; x-ray structural analysis showed them to be amorphous, and IR spectroscopy, to be conjugated polymers. Electrical conductivity was studied at 20-300°C and 1×10^{-4} or 760 mm Hg after degassing at 1×10^{-4} mm Hg and 50°C for 3 hr. All the polymers showed a positive temperature coefficient and an exponential temperature dependence of conductivity. Electrical conductivity at 50°C ranged from 1×10^{-12} to 1×10^{-9} ohm⁻¹·cm⁻¹, and activation energy, from 0.3 to 1.74 ev (no degassing). Study of the effect of surface adsorption on the semiconducting properties of the 1,1'-diacetylferrocene polymer showed that the high activation energies (1.5 ev) are

Card 2/3

ACCESSION NR: AP4030375

caused mostly by surface adsorption and only to a small degree by π-electron excitation from the valence to the conduction band.
Orig. art. has: 4 tables, 2 figures, and 3 formulas.

ASSOCIATION: Moskovskiy institut neftekhimicheskoy i gazovoy
promyshlennosti im. Gubkina (Moscow Institute of the Petrochemical
and Gas Industry)

SUBMITTED: 02Apr63 DATE ACQ: 07May64 ENCL: 00

SUB CODE: CH,PH NO REF Sov: 011 OTHER: 002

Card 3/3

APPENDIX Data on the relationship between the mean number of days per year spent in hospital and the mean number of days per year spent in the community among patients with mental illness.

SQUIFFE AN SSSR Doklady v. 194 p. 4 1964 511-514

ABSTRACT. The electronic structure of the α -form cyclopolymers and the crosslinking reaction was studied in the mass spectrum by measuring the dependence of the loss of the α -form on the temperature of the ion source. The results obtained show that the α -form is formed as the β -form is converted into the γ -form, which is reflected in the and

Card 1/3

L 22200-64
ACCESSION NO. A1

indicating the high molecular weight of polyacrylate ester in the polymeric structure. Insoluble

L 2329C-65
ACCESSION NR: AP5000915

where a_1 refers to linear and a_2 to cross linked fractions of the polymer. Our art has been directed to the preparation of such polymers.

ASSOCIATION FOR THE STUDY OF POLYMERS AND THE RUSSIAN ACADEMY OF CHEMICAL PHYSICS INSTITUTE OF POLYMER PHYSICS, SOVIET ACADEMY OF SCIENCES, 194050, GAZOVA 37, TVER, RUSSIA, 194050, 1990-1991.

V. N.

APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001860110008-6"

"APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001860110008-6

APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001860110008-6"

"APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001860110008-6

In the temperature range 20-25°C

Card 1/2

ACCESSION NR: AFS011054

APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001860110008-6"

"APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001860110008-6

SUB CODE: CC, MT

.....M.....

.....N.....

N.....P.....

.....Q.....

.....R.....

APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001860110008-6"

PAUSHKIN, Ya.M.; VISHNYAKOVA, T.P.; KUBASHEVA, L.P.

Preparation of acetyl cyclopentadienyltricarbonylmanganese.
Zhur. ob. khim. 35 no.9;1682-1684 S '65. (MIRA 19-16)

A L 11824-66 EWT(m)/EWP(j)/T/ETC(m) MW/RM

ACC NR: AP6001493

SOURCE CODE: UR/0191/65/000/012/0010/0012

AUTHOR: Golubeva, I. A.; Vishnyakova, T. P.

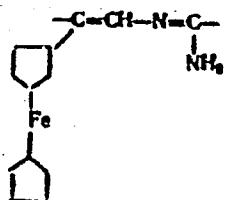
ORG: none

TITLE: Heteropolycondensation of acetylferrocene with urea

SOURCE: Plasticheskiye massy, no. 12, 1965, 10-12

TOPIC TAGS: semiconducting polymer, polycondensation, urea, conjugated polymer, thermal stability, temperature dependence, electric conductivity, organic nitrogen compound, ferrocene

ABSTRACT: A new ferrocene- and nitrogen-containing conjugated polymer.¹⁴¹⁵⁵



has been prepared by heteropolycondensation of acetylferrocene with urea. It is noted that the introduction of ferrocene nuclei into conjugated systems with hetero atoms in the backbone improves thermal stability and produces specific magnetic and electrical properties.¹⁴¹⁵⁵ The reaction was carried out in a metal autoclave in

Card 1/2

UDC: 678.86:66.095.3

L 11824-66

ACC NR: AP6001493

the absence of atmospheric oxygen and in the presence of $ZnCl_2$ catalyst at 110 to 190°C. The polymers were dark infusible powders; the benzene-soluble fraction decomposes at about 350°C and has a mol wt of about 1000. The temperature dependence of electrical conductivity measured in vacuum in the 20—300°C range for degassed samples was exponential in character. Conductivity [at room temperature] was 4.7×10^{-7} mho/cm. Orig. art. has: 1 table and 2 figures. [SM]

SUB CODE: 07.20 / SUBM DATE: none / ORIG REF: 008 / OTH REF: 002 / ATD PRESS: 4478

HW

Card 2/2

L 14204-66 EWP(j)/EWT(m)/T RM

ACC NR: AP6003429

SOURCE CODE: UR/0190/66/008/001/0181/0185

AUTHOR: Vishnyakova, T. P.; Golubeva, I. A.; Paushkin, Ya. M.

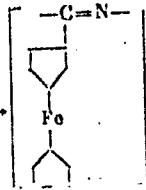
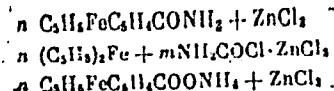
ORG: Moscow Institute of the Petrochemical and Gas Industry im. I. M. Gubkin
(Moskovskiy institut neftekhimicheskoy i gazovoy promyshlennosti)

TITLE: Synthesis of ferrocene-⁷ and nitrogen-containing polymers with a conjugated bond system¹

SOURCE: Vysokomolekulyarnyye soyedineniya, v. 8, no. 1, 1966, 181-185

TOPIC TAGS: organic semiconductor, semiconducting polymer, polynitrile

ABSTRACT: New ferrocene- and nitrogen-containing polymers—polyferrocenylnitriles—have been prepared by polycondensation⁷ of amides or ammonium salts of ferrocenecarboxylic acids. The reaction was conducted in an autoclave in the absence of atmospheric oxygen and in the presence of $ZnCl_2$ catalyst. Polyferrocenylnitrile¹ was prepared at 170—200°C from ferrocenecarboxamide, ammonium ferrocenecarboxylate, as well as from ferrocene proper:



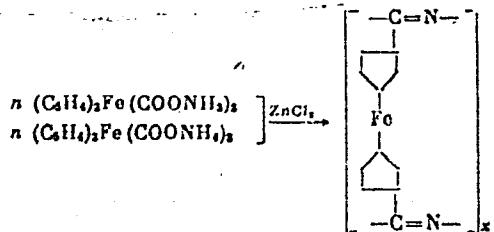
UDC: 541.64+678.86

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L 14204-66

ACC NR: AP6003429

In the case of ferrocenecarboxamide, P_2O_5 and $TlCl_4$ catalysts were used in addition to $ZnCl_2$. The simplest and most effective method was the second (yield, 87% on the ferrocene). Polyferrocenyldinitrile was also prepared at 200°C from 1, 1'-ferrocenedicarboxamide and from diammonium 1, 1'-ferrocenedicarboxylate;



The best method was the second (yield, 44.5% on the ferrocene). All the polymers were black to brown powders; their physical and electrical properties are shown in Table 1. Structures were confirmed by IR spectroscopy. The temperature dependence

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ACC NR: AP6003429

Table 1. Properties of ferrocenylnitriles

	M.p., °C		Mol. wt	N, spin/g	σ_{50} mho/cm	ΔE , ev
	DMF*	DMF soluble				
Polyferrocenyl-nitrile	350—400	>500	1200—1600	10^{17} — 10^{19}	10^{-11} — 10^{-8}	0.724— 0.09
Polyferrocenyl-dinitrile	None	>500	—	10^{18}	10^{-12} — 10^{-14}	0.93— 1.28

*Dimethylformamide

of conductivity of the polymers was exponential in character. Orig. art. has:
 4 tables and 1 figure.

[SM]

SUB CODE: 07/ SUBM DATE: 10Mar65/ ORIG REF: 007/ OTH REF: 002/ ATD PRESS:
 4193

Card 3/3 10

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Copeognatha of the Armenian S.S.R. and methods of their laboratory maintenance. Izv. AN Arm. SSR. Biol. nauki 16 no.9:89-94
(MIRA 17:7)
S*63

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new Late Jurassic cockroach. Paleont. zhur. no. 1:82-87 '64.
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DANILEVSKIY, A.S.; MARTYNOVA, O.M.; NOVOZHILOVYY, N.I.;
PONOMARENKO, A.G.; POPOV, Yu.A.; RODENDORF, B.B.; CHERNOVA,
O.A.; SHAROVYY, A.G.; ORLOV, Yu.A., glav. red.; MARMOVSKIY,
B.P., zam. glav. red.; RUZHENTSEV, V.Ye., zam. glav. red.;
SOKOLOV, B.S., zam. glav. red.; OSIPOVA, L.S., red. izd-va;
MAKUNI, Ye.V., tekhn. red.

[Fundamentals of paleontology; reference book in 15 volumes
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(Arthropoda, Fossil)

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(Cockroaches, Fossil)

VISHNYAKOVA, V.N.

New species of genus Kunguroblattina from the Lower Permian
in the Ural Mountain region. Paleont. zhur. no.4:50-59 '65.
(MIRA 19:1)

l. Paleontologicheskiy institut AN SSSR. Submitted Feb. 27,
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(Peas)

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tekhn. red.

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[The new, tested by life] Novoe, proverennoe zhizn'iu. Mo-
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skoy oblasti(for Kondrat'yev).
(Moscow Province--Stock and stock breeding)

BOYEV, Ivan Dmitriyevich; VISHNYAKOVA, Ya.A., red.; YELAGIN, A.S.,
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[Seven-year plan in four years] Semiletka v chetyre goda.
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1. Direktor sovkhoza "Temishbekskiy" Stavropol'skogo kraya
(for Boyev). (State farms)

SHARSHAVENKOV, Vasiliy Ivanovich, svinar'-mekhanizator; VISHNYAKOVA, Ye.A.,
red.; KLYUCHEVA, T.D., tekhn.red.

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(MIRA 15:5)

1. Sovkhoz "Chkalovskiy" Kalizhskoy oblasti (for Sharshavenkov).
(Swine)

GONCHARENKO, Vsevolod Antonovich; VISHNYAKOVA, Ye.A., red.; POPOV, N.D.,
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1. Direktor sovkhoza "Gazyrskiy" Krasnodarskogo kraya (for Goncharenko)
(Krasnodar Territory—Farm management)

DIANOV, Mikhail Ivanovich, Geroy Sotsialisticheskogo Truda; VISHNYA-
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36 p. (Dlia slushatelei sel'skikh nachal'nykh shkol i kruzh-
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ZENIN, Vasiliy Polikarpovich; VISHNYAKOVA, Ye.A., red.; MARAKASOVA,
L.P., tekhn.red.

[Great work of concern to all] Bol'shoe vsemnarodnoe delo.
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1. Sekretar' Ryazanskogo obkomu Kommunisticheskoy partii Sovetskogo
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(Ryazan Province--Agriculture)

PLATONOV, Grigorij Fedorovich; VISHNYAKOVA, Ye.A., red.; YELAGIN,
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[Over-all mechanization is the foundation of success] Kompleksnaja
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(for Platonov).
(Stock and stockbreeding)

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V.Ye. (Moskva, V-261, Leninskiy prosp. d. 81, kv.87); MURAV'YEVA, N.I.
(Moskva, D-67, Volokolamskoye shosse, d. 80, kv.71); ASONOVA, N.K.
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Instituta eksperimental'noy i klinicheskoy onkologii AMN SSSR
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Instituta eksperimental'noy i klinicheskoy onkologii AMN SSSR
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I.M., inzh.; ANDREYEVA, L.P., inzh.; BISIENKEVICH, G.V., inzh.;
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A.I., otv. red.; YUNITSKIY, V.P., red.; FLIGEL'MAN, S.M., red.;
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92 p.

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USSR/Microbiology. Microbes Pathogenic for Man and Animals F

Abs Jour : Ref Zhur-Biol., No 13, 1958, 57776

Author : Rozina Ts. S., Pedenko A. I., Devanisskaya R. D.,
Vishnyakova Yu. N.

Inst : Kharkov Scientific-Research Institute of Vac-
cines and Sera

Title : Bacteriological Characteristics of Diphtheria
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Orig Pub : Tr. Kharkovsk. n.-i in-ta vaktsin i syvorotok,
1957, 24, 91-98

Abstract : No abstract

Card 1/1

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(Russia—Economic policy)

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LYSENKO, Trofim Denisovich, akademik, agrobiolog; VISHNYAKOVA, Ye.A.,
red.; AVDEYEVA, V.A., tekhn.red.

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34 p.

1. Direktor Instituta genetiki AN SSSR (for Lysenko).
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T.D., tekhn.red.

[Discussions on economic aspects of agriculture on state farms]
Besedy ob ekonomike sovkhoznogo proizvodstva. Moskva, Izd-vo
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Vop.onk. 5 no.5:637-640 '59. (MIRA 12:12)
(TUMORS) (CHEMOTHERAPY)

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1. Iz klinicheskogo otdela Instituta eksperimental'noy patologii i terapii raka AMN SSSR (dir. - chl.-kor. AMN SSSR prof. N.N. Blokhin).

(NITROGEN MUSTARDS, ther. use,
5-(β -chloroethyl)amino-4-cathyl-uracil in myelocytic leukemia & other forms of cancer (Rus))

(URACIL, rel. cpds.
same)

(LEUKEMIA, MYLOCYTIC, ther.
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1. Institut lesa i dravestny Sibirs'kogo otdeleniya AN SSSR,
Krasnoyarsk.

VISHNYASKAYTE, A. I.

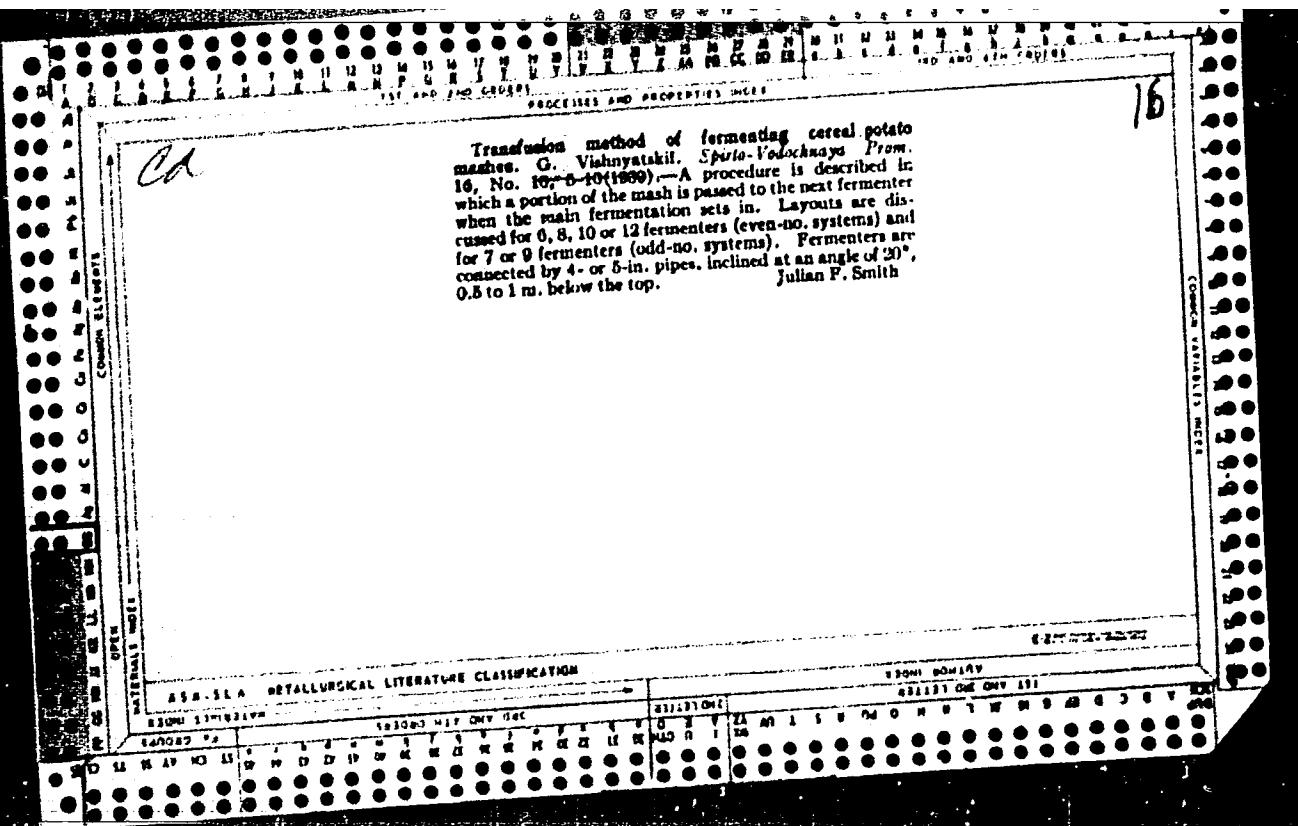
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Moscow, 3-10 Aug 64.

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veter. nauk; VISHNYAVSKAS, A.YU. [Vishnyavskas, A.], starshiy
nauchnyy sofrunkt

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1. Vsesoyuznyy institut gel'vitologii imeni akademika Skryabina
(for Demidov). 2. Pitovskiy nauchno-tekhnicheskiy veterinarnyy
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(Continued on next card)

11036

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USSR/Agricultural Machinery 4302.0500 Aug 1947

"Slats for Reaper Conveyors," Ya. K. Vishom, 1 p

"MTS" No 8

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of layers instead of a single block of wood.

11036

LC

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Nature of the cathodic polarization of cadmium in a cyanide
electrolyte. Trudy AN Lit. SSR. Ser. B no.2:19-30 '62.

1. Institut khimii i khimicheskoy tekhnologii AN Litovskoy SSR.
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Characteristic phenomena of polarization observed during electro-deposition of silver from cyanide electrolytes. Trudy AN Lit. SSR
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1. Institut khimii i khimicheskoy tekhnologii AN litovskoy SSR.

VISHOMIRSKIS, R.M. [Visomirskis, R.]

Cathodic polarization of copper in cyanin electrolyte. Liet ak
darbai B no.4:103-110 '59. (EPAI 9:3)

1. Institut khimii i khimicheskoy tekhnologii AN Litovskoy SSR.
(Copper) (Cyanides) (Electrolytes)
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"Über das wesen der kathodischen polarisarion bei der elektrolytischen
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